

Electrochemical Soil Remediation for Contaminant Removal

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Introduction

Soil is an essential and valuable natural resource. Without healthy soil, life on Earth would be impossible, as 95% of human food comes from the soil. Ensuring soil remains healthy and productive is crucial for human survival. The introduction of materials, biological organisms, or energy into soil can alter its quality and impact groundwater, plant life, and food production. This makes soil pollution a significant global environmental concern. Over time, soils have become contaminated with both organic and inorganic pollutants from various industrial, agricultural, and military activities.

Heavy metals, in particular, pose a substantial threat to the environment and public health. These metals interact with natural soil components, causing both immediate and long-term issues such as delayed hydration reactions and the leaching of metals into groundwater. Lead (Pb), cadmium (Cd), and chromium (Cr) are the most commonly found toxic metals at contaminated sites, posing serious risks such as neurological damage, liver and kidney damage, and cancer.

Addressing soil pollution requires the development of efficient, cost-effective remediation strategies to protect the environment for future generations.

Soil Pollution and its Sources

Soil pollution occurs when harmful substances, such as toxic chemicals, heavy metals, salts, radioactive materials, or disease-causing agents, accumulate in soil and negatively affect plant and animal health. Soil can become polluted in many ways, including:

- Seepage from landfills
- Discharge of industrial waste
- Percolation of contaminated water
- Rupture of underground storage tanks
- Excessive use of pesticides, herbicides, or fertilizers
- Leaching of solid waste

The most common pollutants in soil include petroleum hydrocarbons, heavy metals, pesticides, and solvents.

Soil Remediation Techniques

Soil remediation aims to remove, neutralize, or transform contaminants into harmless substances. There are several approaches to soil remediation, including biological, physical, chemical, and thermal treatments.

1. Biological Remediation

Biological treatment involves using microorganisms to degrade or transform contaminants into non-toxic compounds,

such as carbon dioxide, water, fatty acids, and biomass. It is a low-cost method but often requires more time to achieve complete remediation. Additionally, microbial effectiveness may be limited in highly contaminated soils.

2. Physical and Chemical Treatments

Physical treatments rely on the physical properties of contaminants or soil to remove or contain pollutants. These can involve separation, containment, or chemical alteration of contaminants to reduce their toxicity. Chemical treatments can change the chemical structure of contaminants to make them less harmful. While these methods are generally cost-effective and fast, they may be influenced by factors such as soil composition, including the presence of clay or humic materials.

3. Thermal Treatments

Thermal treatments use heat to break down contaminants, making them volatile or decomposing them. While thermal methods offer quick remediation, they are typically the most expensive, with high energy and equipment costs. The duration of thermal treatment depends on factors like the type of contamination, the size of the polluted area, and soil conditions.

4. Electro Remediation

Electro remediation uses direct electrical current to remove organic and inorganic contaminants, including heavy metals, from the soil. It involves electrochemical processes

that induce chemical reactions in the soil, facilitating the removal of contaminants.

Electro Remediation: Principles and History

Electro remediation, also known as electro kinetic remediation, is a technique where an electric field is applied to the soil to drive the movement of contaminants. This process leverages electroosmosis, electromigration, and electrophoresis to remove pollutants. The method works by applying an electric field across the soil, causing ions and charged particles to move toward the oppositely charged electrode.

- **Electroosmosis:** The bulk movement of water through soil under the influence of an applied electric field. Water moves from the anode to the cathode due to the negative charge on the soil particles.

- **Electromigration:** The movement of ions in the pore fluid toward the electrodes under the electric field.

- **Electrophoresis:** The movement of charged particles and colloids suspended in the pore fluid toward the electrodes.

Electro remediation was first explored in the 1930s when researchers began using electric potentials to remove ions from soil. In the 1980s, the technique gained popularity for its potential to address groundwater contamination in soils with low permeability. Research in electrochemical remediation accelerated in the 1990s, with notable contributions from teams in the Netherlands,

the United States, and other countries. Today, electrokinetic remediation remains an area of active research, with numerous studies published on its effectiveness.

Factors Affecting Electro Remediation

The efficiency of electro remediation depends on several factors:

- **Electrodes:** Various electrode materials are used, with graphite being the most common due to its cost-effectiveness. However, other materials, such as lead dioxide (PbO₂) or boron-doped diamond, may offer better performance but come at a higher cost.
- **Spacing:** The distance between electrodes is typically 3 meters, though this can vary depending on the level of contamination.
- **Concentration of Contaminants:** The concentration of pollutants affects the efficiency of electrokinetic processes. At low concentrations, the current primarily flows through clay surfaces, while at higher concentrations, larger anions carry the current, enhancing fluid migration.
- **Cation Exchange Capacity (CEC):** Soils with high CEC require more energy to lower the pH and facilitate the removal of contaminants. Pre-acidification of soil may be necessary in such cases.
- **Anion Retention Capacity:** Soils with high anion retention have a higher net flow of water toward the cathode.
- **Duration of Treatment:** The length of time required for electro remediation is site-

specific, depending on factors like the size of the polluted area and the concentration of contaminants.

Experimental Setup for Electro Remediation

A typical experimental setup for electro remediation includes the following components:

- **Electrochemical Cell:** A container holding the soil sample during the tests.
- **Electrodes:** A pair of stainless steel electrodes, typically inserted into the soil to create the electric field.
- **Power Supply:** A stabilized direct current generator, connected to the electrodes via copper wires.
- **Pore Fluid Collection Tanks:** Containers for collecting fluids that migrate due to electroosmotic flow.
- **Gas Vents:** Valved vents to release gases produced during electrochemical reactions, such as oxygen at the anode and hydrogen at the cathode.

Conclusion

Electro remediation offers a promising solution for cleaning contaminated soils, particularly those polluted with heavy metals. By applying an electric field, this technique can facilitate the removal or neutralization of harmful contaminants, providing an effective and sustainable approach to soil remediation. As research in this field progresses,

electrokinetic methods may become a more widely adopted solution for addressing soil pollution worldwide.

Reference

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